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## Chloro(5,10,15,20-tetraphenylporphyrinato)manganese(III) with $4/m$ Symmetry

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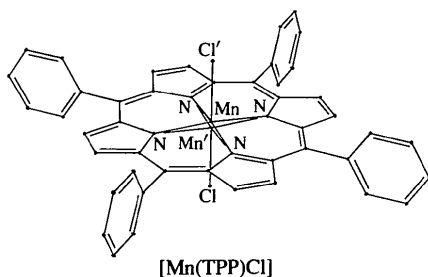
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### Abstract

The crystal structure of  $[\text{Mn}(\text{TPP})\text{Cl}]$  in space group  $I4/m$  (where TPP is  $\text{C}_{44}\text{H}_{28}\text{N}_4$ ) has been determined. The unit cell contains two full molecules, with one eighth of a molecule unique. An out-of-plane model for the Mn atom was applied and all non-H atoms were refined anisotropically. The Mn—N distance is 2.002 (3) Å, the axial Mn—Cl distance is 2.297 (15) Å and the out-of-plane displacement of the Mn atom is 0.16 Å. The possibility of a reverse-doming porphyrin core conformation is mentioned briefly.

### Comment

For tetragonal  $[\text{Mn}(\text{TPP})\text{Cl}]$  with required  $4/m$  symmetry and a five-coordinate manganese(III) ion, the Mn and Cl atoms are required to be disordered along the axis at  $0,0,\pm z$ . The coordination group distances that result are Mn—N = 2.002 (3),  $\text{Ct}\cdots\text{Cl} = 2.452$  (3), Mn—Cl = 2.297 (15) Å, and a Mn-atom displacement of  $\pm 0.16$  Å.



These values are smaller than expected for a five-coordinate  $\text{Mn}^{\text{III}}$  porphyrinate. Indeed, in the crystal structure of  $[\text{Mn}(\text{TPP})\text{Cl}]$  as the acetone solvate with no required symmetry for the molecule (Tulinsky & Chen, 1977), an axial Mn—Cl bond distance of 2.363 (3) Å, an average Mn—N bond length of 2.008 (15) Å and a  $\text{Ct}\cdots\text{Cl}$  distance of 2.633 Å were observed; the Mn-atom displacement from the mean plane of the four N atoms was 0.27 Å. Essentially identical values were obtained for the chloroform solvate of  $[\text{Mn}(\text{TPP})\text{Cl}]$  (Hoard, 1975). We believe that the coordination group parameters for the tetragonal form of  $[\text{Mn}(\text{TPP})\text{Cl}]$  are underestimated and that this underestimation results

from a reverse doming of the porphyrinato core that is caused by crystal packing and obscured by the crystallographically required  $4/m$  symmetry. We note that a similar phenomenon is also observed in the structures of two crystal forms of  $[\text{Fe}(\text{TPP})\text{Cl}]$  and  $[\text{Zn}(\text{TPP})(\text{H}_2\text{O})]$ . In the monoclinic form of  $[\text{Fe}(\text{TPP})\text{Cl}]$  with no required symmetry, the axial Fe—Cl bond length is 2.211 (1) Å, the average Fe—N distance is 2.070 (9) Å and the Fe-atom displacement is 0.49 Å from the plane of the four N atoms (Scheidt & Finnegan, 1989). However, a set of significantly smaller values [2.192 (12), 2.049 (9) and 0.38 Å, respectively] were reported for the tetragonal structure with required  $4/m$  symmetry (Hoard, Cohen & Glick, 1967). In addition, in the two Zn complexes, the tetragonal form (Glick, Cohen & Hoard, 1967; Golder, Povey, Silver & Jassim, 1990) has smaller coordination group parameters than the triclinic form (Cheng & Scheidt, 1995), although the comparison is made more difficult by the fact that the Zn—O( $\text{H}_2\text{O}$ ) bond is affected by strong hydrogen bonding to the aqua ligand in the triclinic form.

We note that there are many example of five-coordinate  $\text{MTPP}(\text{X})$  complexes which crystallize in the space group  $I4/m$ ; see Table XXVI of Scheidt & Lee (1987) for a number of examples.

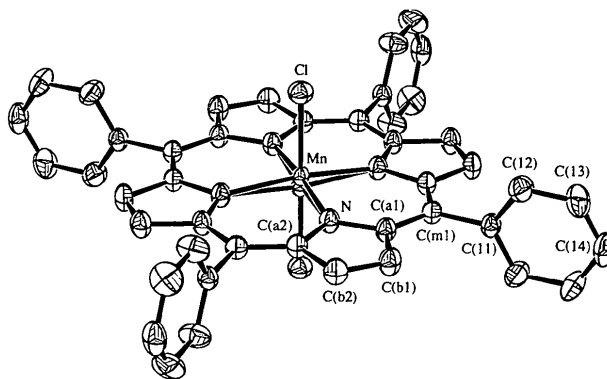


Fig. 1. ORTEP diagram of  $[\text{Mn}(\text{TPP})\text{Cl}]$  showing the labeling scheme. Displacement ellipsoids are drawn at the 30% probability level. Porphyrin H atoms are omitted for clarity.

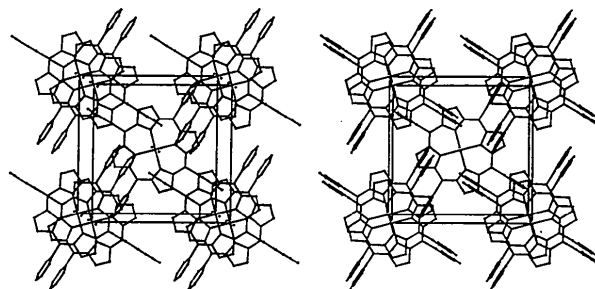


Fig. 2. Stereoscopic packing diagram of  $[\text{Mn}(\text{TPP})\text{Cl}]$ .

## Experimental

This new crystal form with required  $4/m$  symmetry for the title compound was obtained in the course of our synthetic work on  $\mu$ -oxo- and  $\mu$ -hydroxomanganese(III) tetraphenylporphyrinate complexes, and recrystallized from toluene and hexanes.

### Crystal data

[Mn(C <sub>44</sub> H <sub>28</sub> N <sub>4</sub> )Cl]	Mo $K\alpha$ radiation
$M_r = 703.09$	$\lambda = 0.71073 \text{ \AA}$
Tetragonal	Cell parameters from 246
$4/m$	reflections
$a = 13.463 (3) \text{ \AA}$	$\theta = 5.1\text{--}20.8^\circ$
$c = 9.864 (4) \text{ \AA}$	$\mu = 0.481 \text{ mm}^{-1}$
$V = 1787.9 (9) \text{ \AA}^3$	$T = 293 (2) \text{ K}$
$Z = 2$	Rectangular prism
$D_x = 1.306 \text{ Mg m}^{-3}$	$0.35 \times 0.25 \times 0.10 \text{ mm}$
$D_m$ not measured	Black

### Data collection

Enraf–Nonius FAST area-detector diffractometer	771 observed reflections
Ellipsoid-mask fitting	$[I > 2\sigma(I)]$
Absorption correction: none	$R_{\text{int}} = 0.0603$
4391 measured reflections	$\theta_{\text{max}} = 26.62^\circ$
946 independent reflections	$h = -15 \rightarrow 15$
	$k = -16 \rightarrow 16$
	$l = -12 \rightarrow 7$

### Refinement

Refinement on $F^2$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$R(F) = 0.0546$	$\Delta\rho_{\text{max}} = 0.306 \text{ e \AA}^{-3}$
$wR(F^2) = 0.1293$	$\Delta\rho_{\text{min}} = -0.348 \text{ e \AA}^{-3}$
$S = 1.084$	Extinction correction: none
946 reflections	Atomic scattering factors
73 parameters	from <i>International Tables</i>
H atoms idealized, riding	for <i>Crystallography</i> (1992,
(C—H 0.93 \AA)	Vol. C, Tables 4.2.6.8 and
$w = 1/[\sigma^2(F_o^2) + (0.0641P)^2$	6.1.1.4)
$+ 2.7218P]$	
where $P = (F_o^2 + 2F_c^2)/3$	

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

$$U_{\text{eq}} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^*a_i \cdot a_j$$

	$x$	$y$	$z$	$U_{\text{eq}}$
Mn	0	0	0.0158 (15)	0.034 (2)
Cl	0	0	0.2486 (3)	0.0551 (8)
N	-0.0379 (2)	0.1433 (2)	0	0.0435 (8)
C(a1)	0.0249 (3)	0.2240 (3)	0	0.0443 (9)
C(a2)	-0.1329 (3)	0.1822 (3)	0	0.0429 (9)
C(b1)	-0.0316 (3)	0.3134 (3)	0	0.0546 (11)
C(b2)	-0.1273 (3)	0.2882 (3)	0	0.0526 (11)
C(m1)	0.1279 (3)	0.2203 (3)	0	0.0431 (9)
C(11)	0.1841 (3)	0.3155 (3)	0	0.0434 (9)
C(12)	0.2109 (3)	0.3602 (3)	-0.1190 (4)	0.0673 (10)
C(13)	0.2655 (3)	0.4474 (3)	-0.1179 (5)	0.0786 (12)
C(14)	0.2935 (3)	0.4898 (3)	0	0.0694 (15)

Table 2. Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

Mn—Mn <sup>i</sup>	0.31 (3)	C(a2)—C(m1 <sup>ii</sup> )	1.385 (5)
Mn—N	2.002 (3)	C(a2)—C(b2)	1.429 (5)
Mn—Cl	2.297 (15)	C(b1)—C(b2)	1.332 (6)
Mn—Cl <sup>i</sup>	2.609 (15)	C(m1)—C(11)	1.488 (5)

N—C(a1)	1.377 (5)	C(11)—C(12)	1.368 (4)
iv—C(a2)	1.381 (5)	C(12)—C(13)	1.384 (5)
C(a1)—C(m1)	1.387 (5)	C(13)—C(14)	1.350 (5)
C(a1)—C(b1)	1.424 (5)		
Mn <sup>i</sup> —Mn—N	85.5 (4)	N—C(a2)—C(b2)	109.2 (3)
N <sup>ii</sup> —Mn—N	89.65 (6)	C(m1 <sup>iii</sup> )—C(a2)—C(b2)	124.9 (4)
N <sup>i</sup> —Mn—N	171.1 (8)	C(b2)—C(b1)—C(a1)	107.6 (3)
N—Mn—Cl	94.5 (4)	C(b1)—C(b2)—C(a2)	107.8 (3)
C(a1)—N—C(a2)	105.7 (3)	C(a2 <sup>iii</sup> )—C(m1)—C(a1)	123.9 (3)
C(a1)—N—Mn	127.1 (2)	C(a1)—C(m1)—C(11)	118.5 (3)
C(a2)—N—Mn	126.9 (2)	C(a2 <sup>iii</sup> )—C(m1)—C(11)	117.6 (3)
Mn—N—Mn <sup>i</sup>	8.9 (8)	C(12 <sup>iv</sup> )—C(11)—C(12)	118.2 (4)
N—C(a1)—C(m1)	125.9 (3)	C(12)—C(11)—C(m1)	120.9 (2)
N—C(a1)—C(b1)	109.7 (3)	C(11)—C(12)—C(13)	120.5 (3)
C(m1)—C(a1)—C(b1)	124.4 (4)	C(14)—C(13)—C(12)	120.9 (4)
N—C(a2)—C(m1 <sup>iii</sup> )	125.9 (3)	C(13)—C(14)—C(13 <sup>iv</sup> )	119.0 (4)

Symmetry codes: (i)  $-x, -y, -z$ ; (ii)  $-y, x, z$ ; (iii)  $y, -x, -z$ ; (iv)  $x, y, -z$ .

Cell determination and intensity data collection were performed on an Enraf–Nonius FAST area-detector diffractometer with an Mo rotating-anode source. Our detailed methods and procedures for small-molecule X-ray data collection with the FAST system have been described previously (Scheidt & Turowska-Tyrk, 1994). The centrosymmetric space group  $I4/m$  was chosen based on the  $E$  statistics and consequently proved by the successful refinement.

Data collection: *MADNES* (Messerschmitt & Pflugrath, 1987). Cell refinement: *MADNES*. Data reduction: *MADNES*. Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993). Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *SHELXL93*.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: BK1136). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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